

(An ISO 3297: 2007 Certified Organization) Vol. 4, Issue 4, April 2015

Growth and Structural Characterization of DVT Grown SnSeRe_X(X = 0, 0.1, 0.2, 0.3, 0.4) Crystals

Trupti Patel¹, G.K. Solanki², Vimal S. Joshi ^{3*}

Assistant Professor in Physics, Department of General, Government Engineering College, Bharuch, Gujarat, India¹

Reader, Department of Physics, Sardar Patel University, Vallabh Vidyanagar, Gujarat, India²

Principal, Shri R. K. Parikh Arts and Science College, Petlad, Gujarat, India.³

ABSTRACT: Among the IV-VI semiconductor compounds, Tin sulphide (SnS), Tin selenide (SnSe), Germanium Sulphide (GeS) and Germanium Selenide (GeSe) have the layered orthorhombic structure with eight atoms per unit cell forming biplanar layers normal to the largest c axis [1-4]. In the present work, investigation on growth of single crystals of SnSeRex (x = 0, 0.1, 0.2, 0.3, 0.4) by direct vapour transport technique (DVT) using two zone horizontal furnace is reported. Confirmation of stoichiometric proportion of constituent elements and determination of crystal structure of grown crystals is done by Energy Dispersive Analysis of X-rays (EDAX). The structural behaviour was accomplished by X-Ray diffraction (XRD) studies. As a result, it seen to be orthorhombic structure and results are discussed.

KEYWORDS: Semiconducting IV-VI compounds, growth from vapour, EDAX and XRD analysis

I. INTRODUCTION

During recent years, science and technology enjoy a higher degree of sophistication due to the availability of high quality materials. The layered compounds with a distorted rock salt structure of the orthorhombic space group; among the binary IV-VI semiconductor compounds are tin sulphide (SnS), tin selenide (SnSe), germanium sulphide (GeS) and germanium selenide (GeSe) [1,5,6]. These compounds are also known as lamellar semiconductors which attracted considerable attention because of their important specific properties in the field of optoelectronics [7], holographic recording systems [8], electronic switching [9-10] and infrared production and detection. Moreover SnSe is a semiconductor with a band gap of about 1 eV having a potential as an efficient solar cell material [11-12]. It is clear from that the literature survey that a lot of research work has been done on Tin Selenide single crystal; however, the work on compounds Tin Selenide with substitution of transition metal is almost negligible. Hence, it was decided to work on SnSeRex (x = 0, 0.1, 0.2, 0.3, 0.4) compounds by direct vapour transport technique (DVT) using two zone horizontal furnace. The results are described in the paper.

II. EXPERIMENTAL TECHNIQUE

CRYSTAL GROWTH

In present investigations, crystal growth of SnSeRe family was done by direct vapour transport technique using dual zone horizontal furnace. The furnace was constructed by University Science and Instrumentation Centre (USIC), Sardar Patel University by using a special sillimanite threaded tube closed at one end, 450 mm in length, 70 mm outer diameter, 56 mm inner diameter with threaded pitch of 3 mm, imported from koppers Fabriken Feuerfester, Germany. The furnace was controlled by a microcontroller with thermal profile programming facility. High quality quartz ampoules were used for crystal growth which having dimensions of 24 cm length, 2.4 cm outer diameter and 2.2 cm inner diameter. Firstly boiled water was used to wash the ampoule after then with hot mixture of concentrated HNO₃ and H_2So_4 taken in equal proportion, followed by washing of ampoule with distilled water. Then ampoule was filled with concentrated HF and heated until whole HF evaporated, so that the inner surface of the ampoule become rough which enhances the nucleation



(An ISO 3297: 2007 Certified Organization)

Vol. 4, Issue 4, April 2015

process during the growth. At the end this ampoule was washed by double distilled water at least 8 to 10 times. A cleaned ampoule was kept in oven for nearly 24 hour to make it moisture free.

Highly pure powder of Sn (99.99%), Se (99.99%) and Re (99.99%) were taken in a stoichiometric proportion in the ampoule for the compound preparation. It was evacuated to a 10^{-5} Torr and then sealed. The sufficient care was taken for vigorous shaking so as to distribute the mixture along the length uniformly. The ampoule was set in a horizontal furnace. Its temperature was slowly raised till it reaches to 1073 K. It was maintained at this temperature for a period of 3 day for charge preparation. The ampoule was then slowly cooled and brought to room temperature. The resulting free flowing shiny homogeneous polycrystalline powder was achieved. This charged ampoule was placed in a dual zone horizontal furnace. For different member of the series the temperatures of hot zone and cold zone of the ampoule were kept as shown in Table 1. Finally resulting crystals were collected from the ampoule. The appropriate growth conditions for all samples are reported in Table 1.

Crowth parameters	SnSeRex (X= 0, 0.1, 0.2, 0.3, 0.4)					
Growin parameters	<i>x=0</i>	<i>x=0.1</i>	x=0.2	x=0.3	<i>x=0.4</i>	
Source zone temperature (°C)	620	610	600	590	580	
Growth zone temperature(°C)	600	590	580	570	560	
Ampoule dimensions	23cm length and 2.2 cm inner diameter					
Vacuum level (torr)	10-5					
Growth period (Day)	7					
Heating Rate	30 (°C/hour)					
Cooling Rate	20 (°C/hour)					
Average Dimensions $(cm \times cm)$	0.9 × 0.6	0.7×0.5	0.9 imes 0.5	0.6 ×0.4	0.9 imes 0.5	

Table 1: Growth Conditions For SnSeRex (X = 0, 0.1, 0.2, 0.3, 0.4) Single Crystals by DVT

STRUCTURAL CHARACTERIZATION

In present case, the grown crystals have been characterized for their structural propertied using EDAX and XRD techniques. The chemical proportions of elements in grown crystals are obtained from the Energy Dispersive Analysis of X-rays (EDAX) which has been done at kalabhavan, M. S. University, Baroda. The energy dispersive analysis using X-rays (EDAX) experiment has been carried out on each grown sample for the determination of the stoichiometric proportion of Sn, Se and Re. The X-ray diffractograms were obtained for each sample with Philips X-ray diffractometer PW 1820 employing CuK_a radiation. The powder obtained during the growth process for each sample was prepared for the X-ray diffraction study experiment. The lattice parameters, hkl reflections and d spacing of the grown crystals were determined by comparison with JCPDS data as well as by using power X software. The X-ray diffraction patterns obtained for SnSeRex (x = 0, 0.1, 0.2, 0.3, 0.4) are shown in Fig. 1 (a), (b), (c), (d) and (e) respectively.



(An ISO 3297: 2007 Certified Organization) Vol. 4, Issue 4, April 2015







Fig. 1(c): X-ray Diffractogram of SnSeRe_{0.2} (DVT) crystal



Fig. 1(b): X-ray Diffractogram of SnSeRe_{0.1} (DVT) crystal



Fig. 1(d): X-ray Diffractogram of SnSeRe_{0.3} (DVT) crystal



Fig. 1(e): X-ray Diffractogram of SnSeRe_{0.4} (DVT) crystal



(An ISO 3297: 2007 Certified Organization)

Vol. 4, Issue 4, April 2015

III. RESULTS AND DISCUSSIONS

The single crystals of SnSeRex (x = 0, 0.1, 0.2, 0.3, 0.4) were grown using DVT technique. The weight of elements taken for growth is nearly equal to the weight of elements from EDAX analysis is shown in Table 2. It confirms that the stoichiometry of the each grown crystal has been maintained. Table 2: Results Obtained From EDAX spectra

Samples	Wt (%) of the elements taken for crystal growth			Wt (%) of the elements obtained from EDAX		
	Sn	Se	Re	Sn	Se	Re
SnSe	60.05	39.95	-	56.57	43.43	-
SnSeRe _{0.1}	54.88	36.5	8.06	60.16	37.81	2.03
SnSeRe _{0.2}	50.53	33.61	15.85	52.39	34.31	13.3
SnSeRe _{0.3}	46.41	31.22	22.37	41.96	30.93	23.2
SnSeRe _{0.4}	43.51	29.13	27.35	38.85	31.11	30.1

The X-ray diffractograms of mentioned grown crystals by DVT technique were analysed. It was noted from Fig.1 that the diffraction pattern of SnSeRex (x = 0, 0.1, 0.2, 0.3, 0.4), shown a marked resemblance with that of binary compound SnSe. The sharp peaks indicate the good crystalline structure of grown compounds. Thus these mixed compounds can also be indexed on an orthorhombic unit cell basis. The values of the different lattice parameters obtained from the analysis of the diffractograms of each crystal are presented systematically in Table 3.

 Table 3: Results Obtained from X- ray
 Diffractogram

Samples	La	Unit cell Density		
	a (Å)	b (Å)	c (Å)	(g/cm^3)
SnSe	11.42	4.19	4.46	2.24
SnSeRe _{0.1}	11.48	4.156	4.431	2.47
SnSeRe _{0.2}	11.476	4.14	4.452	2.68
SnSeRe _{0.3}	11.476	4.148	4.454	2.89
SnSeRe _{0.4}	11.452	4.139	4.458	3.11

The

observations on the values of lattice parameters, unit cell volume and density are presented in Table 3. It indicates that as the stoichiometric proportion of rhenium increases the unit cell density increases since the total weight of the atoms in the unit cell is also decreases. The lattice parameters a; b and c obtained for SnSe single crystals were compared and found in agreement with those earlier reported by different investigators [13-14].

ACKNOWLEDGEMENT

The authors are thankful to Department of Physics, Sardar Patel University, Vallabh Vidyanagar and M. S. University of Baroda for carrying out some part of this work. Corresponding author Vimal Joshi is thankful to the University Grants Commission, New Delhi for Financial Assistance under Minor Research Project Scheme. The authors are also grateful to Prof. U. H. Patel for their valued guidance and support. The authors are also thankful to Petlad Education Trust for motivation.



(An ISO 3297: 2007 Certified Organization)

Vol. 4, Issue 4, April 2015

REFERENCES

- [1] A. Okazaki, J. Phys. Soc. Japan, Vol.13, pp.1151, 1958
- [2] W. Hofmann, Z. Kristallogr. Vol. 91, pp.61, 1935
- [3] A. Okazaki and I. Uada, J. Phys. Soc. Jpn, Vol.11, pp.470, 1956
- [4] Okazaki, J. Phys. Soc. Japan, Vol.13, pp.1151, 1958
- [5] S. Logothetidis and H.M. Polatoglou, Phys. Rev., Vol. B36, pp.7491, 1987
- [6] G. Valiukonis, D.A. Gussienova, G. Krivaite and A. Sileika, Phys. Stat. Solidi (b), Vol.135, pp. 299, 1986
- [7] M. Parentau and M. Carlone, Phys. Rev., Vol.B41, pp.5227, 1990
- [8] J.P. Singh and R.K. Bedi, J. Appl. Phys., Vol.68, pp.2776, 1990
- [9] J.J Lofersky, J.Appl. Phys., Vol.27, pp.777,1956
- [10] M. Parentau and M. Carlone, Phys. Rev., Vol.B41,pp. 5227, 1990
- [11] M. Rodoat, Acta Electronica, Vol.18, pp.345, 1975
- [12] T.H. Patel, Rajiv Vaidya, S.G. Patel, Journal of Crystal Growth, Vol.253, pp.52–58, 2003
- [13] Chattopadhyay, T., Pannetier, J. & Vonschnering, J. Phys. Chem. Solids, Vol. 47, pp.879-885, 1986
- [14] Baumgardner, W. J., Choi, J. J., Lim, Y.-F. & Hanrath, J.Am.Chem.Soc., Vol. 132, pp.9519-95, 2010