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# Investigation on the Properties of Zinc Thiourea Sulphate Crystal for NLO Application

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**ABSTRACT**: A semi-organic non-linear optical single crystal of zinc thiourea sulphate has been grown by slow evaporation solution growth technique. The single crystal XRD analysis confirms that the crystal belongs to the orthorhombic system with non-centrosymmetric space group Pmmm. Powder X-ray diffraction analysis was carried out for the grown crystal in the range 20°-80°. The presence of functional groups was confirmed by using FTIR spectral analysis. The UV-Vis spectrum shows that it has a good transmittance in the entire visible region with the lower cut-off wavelength at 253 nm. The thermal characteristics of zinc thiourea sulphate were analyzed by thermogravimetric (TGA) and differential thermal analysis (DTA). The mechanical behavior of the grown crystal was studied using Vickers microhardness tester. Nonlinear optical characteristics of zinc thiourea sulphate have been studied using Q-switched Nd:YAG laser ( $\lambda$ =1064 nm). The second harmonic generation conversion efficiency of zinc thiourea sulphate is 0.8 times that of standard KDP crystal.

KEYWORDS: Nonlinear optical, crystal growth, mechanical properties, thermal analysis

## I. INTRODUCTION

The nonlinear optical properties of the organic molecular crystals have received a great deal of interest for the past two decades due to their extensive application in the fields like laser technology, telecommunication, optical information processing and storage [1-2].

The semi-organic materials have the potential for combining the high optical nonlinearity and flexibility of organics with temporal, thermal stability and excellent transmittance of inorganics [3-7]. Zinc thiourea sulfate (ZTS) is an efficient semiorganic nonlinear material for type II second harmonic generation [8-10]. Zinc Thiourea Sulfate having molecular formula  $Zn(NH_2CSNH_2)_3SO_4$  is one such a novel promising non-linear optical metal organic crystal having essential applications in the area of second harmonic generation device as well as laser tuned experiments [11]. While, a centro symmetric Thiourea molecule combines with inorganic material, it turns into non centro symmetric (NLO) complexes [12] which include both organic and inorganic advantages which known as semi organic material. This makes ZTS as a good candidate for non linear optical applications having 0.8 times more SHG efficiency than KDP crystal. The growth and various studies of doped and undoped ZTS crystals have been reported in a number of publications [13-17]. In the present work, zinc thiourea sulphate has been grown by slow evaporation technique and investigated the optical, thermal, and mechanical properties through various characterization studies such as X-ray diffraction, UV-Vis, FTIR and TGA and DTA.

## **II. MATERIALS AND METHODS**

Single crystal of zinc thiourea sulphate was grown by slow evaporation solution technique. The high purity zinc sulphate heptahydrate (ZnSO<sub>4</sub>.7H<sub>2</sub>O) and thiourea (CS[(NH<sub>2</sub>)<sub>2</sub>]) were taken in the molar ratio of 1:3 and were dissolved in double distilled water. The complex compound forms as follows: ZnSO<sub>4</sub>.7H<sub>2</sub>O+ 3CS (NH<sub>2</sub>)<sub>2</sub>  $\rightarrow$  Zn [CS (NH<sub>2</sub>)<sub>2</sub>] <sub>3</sub>.SO<sub>4</sub>



(An ISO 3297: 2007 Certified Organization)

#### Vol. 5, Issue 11, November 2016

Care was taken to avoid decomposition by keeping the temperature fixed at 50°C during stirring. The prepared homogeneous solution was filtered and kept for evaporation. Transparent colorless ZTS crystals were harvested in around 15 days.

#### III. RESULT AND DISCUSSION

SINGLE CRYSTAL AND POWDER X-RAY DIFFRACTION The obtained XRD data of zinc thiourea sulphate crystal are a=7.795(5) Å, b=11.093(3) Å, c=15.467(6) Å,  $\alpha = \beta = \gamma = 90^{\circ}$  and cell volume=1337.4(2) Å<sup>3</sup>. The single XRD data of zinc thiourea sulphate crystal indicate that the grown crystal belongs to orthorhombic crystal system with P<sub>mmm</sub> space group.



Fig.1. The standard and observed indexed Powder XRD Pattern of the grown ZTS Crystal

Powder X-ray powder diffraction (XRD) is vital for confirming the identity of a solid material and determining crystallinity and phase purity. Fig.1 shows the x-ray powder diffraction pattern of ZTS single crystal. The obtained (hkl) values were indexed using the JCPDS (No.53-0418) software. The sharp well defined Bragg's peaks at specific 2 theta angles found in the spectra show good crystallinity of the grown single crystal. The absorbed prominent peaks of ZTS are (100),(113),(105),(314) and (042). The very high intensity and low FWHM values confirms the good crystalline nature of the grown crystal. The peak positions are well matched with standard diffraction files (JCPDS No.53-0418) of ZTS.

#### FTIR SPECTRAL STUDY

The FTIR spectrum of ZTS is shown in Fig 2. In ZTS compound, there are three thiourea groups and one sulfate ion. Each thiourea group consists of one carbon atom bonding to one sulfur and two nitrogen atoms. Each of the nitrogen atoms in thiourea is connected to two hydrogen atoms. Zinc ion is tetrahedrally coordinated to three sulfur atoms of thiourea and to an oxygen atom of sulfate ion.



(An ISO 3297: 2007 Certified Organization)

Vol. 5, Issue 11, November 2016



Fig.2. FTIR Spectrum of the grown ZTS Crystal

The absorption band observed at  $1633 \text{ cm}^{-1}$  in the spectrum of ZTS corresponds to that of thiourea of about the same frequency  $1625 \text{ cm}^{-1}$  and can be assigned to the NH<sub>2</sub> bending vibration. The absorption observed at  $1519 \text{ cm}^{-1}$  and 940 cm<sup>-1</sup> in the spectrum of ZTS corresponds to the 1470 and 910 cm<sup>-1</sup> absorption of thiourea, respectively and can be assigned to the N-C-N stretching vibration. The increase in the frequency can be attributed to the greater double bond character of the carbon to nitrogen bond on complex formation. The absorption band at about 1402 cm<sup>-1</sup> corresponds to 1417 cm<sup>-1</sup> of thiourea and can be assigned to the C=S stretching vibration. On coordination through sulfur, nature of vibration is slightly changed. The sharp and intense absorption of ZTS at 713 cm<sup>-1</sup> corresponds to the 730 cm<sup>-1</sup> absorption of thiourea. The lowering of frequency can be attributed to the reduced double bond character of the C=S bond on coordination. The absorption indicate the presence of sulfur to metal bonds in ZTS.

## UV-VIS SPECTRAL ANALYSIS

The optical transmission spectrum of ZTS single crystal was recorded in the wavelength region 200-1000 nm and it is shown in Fig.3. For optical fabrications, the crystal should be highly transparent in the considered region of wavelength. The favorable transmittance of the crystal in the entire visible region suggests its suitability for second harmonic generation. The UV absorption edge for the grown crystal was observed to be around 253nm. The lower cut off wavelength is found to be 253nm for ZTS which infinity good agreement with reported value [20].



The optical absorption coefficient ( $\alpha$ ) was calculated from transmittance using the following relation [18]

 $\alpha = \frac{1}{d} \ln \left( \frac{1}{T} \right)$ 

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## Vol. 5, Issue 11, November 2016

where, T is the transmittance and *d* is the thickness of the crystal. As a direct band gap material, the crystal under study has an absorption coefficient ( $\alpha$ ) obeying the following relation for high photon energies (hv)[19].

$$\alpha = \frac{A[hv - E_g]^{\frac{1}{2}}}{hv}$$

where,  $E_g$  is the optical band gap of the crystal and A is a constant. A plot of variation of  $(\alpha h\nu)^2$  versus hv is shown in Fig.4.



Fig.4. Variation of photon energy (hv) with  $(\alpha hv)^2$  of the grown ZTS crystal

 $E_g$  is evaluated using the extrapolation of the linear part [21]. Using Tauc's plot, the energy gap  $E_g$  was calculated as 4.2eV and the large band gap clearly indicates the wide transparency of the crystal. This high band gap value indicates that the grown crystal possesses dielectric behavior to induce polarization when powerful radiation is incident on the material.

#### THERMAL ANALSIS

The thermal analysis are used to find out the weight loss (TGA), melting and decomposition point (DTA) of the grown ZTS crystal. The TG/DTA curves were shown in Fig.5. There occurs no weight loss up to first x-axis ( $250^{\circ}$ C) in the TG curve. This indicates that there is no inclusion of water in the crystal lattice. A major weight loss observed at  $250^{\circ}$ C in the TG curve is assigned to the decomposition of the title compound. During this transition mass of sample reduces by 51%. The first phase of weight loss begins at 223.04C which is attributed to the decomposition as well as the evaporation of the sample. The second phase of weight loss at 312.22°C is may be due to the degradation of the sample itself.



(An ISO 3297: 2007 Certified Organization)

Vol. 5, Issue 11, November 2016



Fig.5. TGA/DTA of curve grown ZTS crystal

DTA curve shows that pure ZTS melts at 231°C which was also confirmed by earlier literature. In the present work, DTA curve shows that ZTS starts to soften at 231.7 °C and it undergoes endothermic transition (peaks) at 250.86°C, 205.3°C and 359.6°C. Also, the plot shows no endothermic or exothermic transitions below 250°C indicating the absence of other phase transitions before the melting point. The sharpness of endothermic peaks in DTA curve shows good degree of crystallinity of the grown crystals. No decomposition up to its melting point at tests the stability of this material for application in lasers, where the crystals are required to withstand high temperatures.

## MICROHARDNESS

Microhardness is the measure of strength of the material and is an important mechanical property of the optical materials in the formation of the fabrication of devices. Physically hardness is the resistance offered by the material to localized plastic deformation caused by scratching or indentation. Vickers microhardness studies were carried out on ZTS crystal by applying load of differed magnitudes. The indentation time was fixed constant for each trial. Repeated trials were performed to ascertain the correctness of the observed results.



Fig.6. Plot of load versus hardness number

The Vickers microhardness number  $H_v$  was calculated using the relation,  $H_v=1.8544 (P/d^2) \text{ kg/mm}^2$ 

where,  $H_v = 1.8$ 

P is the load in kg and d is the diagonal length of indentation impression in mm. The plot of load (P) versus Vickers hardness number  $(H_v)$  is shown in Fig.6. It is observed that the hardness number increases with increasing load. The



(An ISO 3297: 2007 Certified Organization)

#### Vol. 5, Issue 11, November 2016

increase in hardness with the load could be attributed to the heaping up of materials at the edges of impression made by the diamond pyramid.

The hardness value of ZTS crystal is found to 68.9 kg/mm<sup>2</sup> for a load of 100g which confirms the good mechanical stability of the crystal. The relation between load and size of the indentation is given by Meyer's law:  $P = k_1 d^n$  where,  $k_1$  is a constant for a given material and n is the work hardening coefficient. Fig.7. shows the least square fit plot of log (d) versus log (p).



Fig.7. Plot of log d versus log P

The slope of the straight line gives the work hardening coefficient (n). According to Onitsch, n should lie between 1 and 1.6 for hard materials and should be greater than 1.6 for soft materials. Since n is 2.6 for ZTS, it is concluded that the ZTS crystal is a soft material and hardness number increases with the load and it is useful for nonlinear applications.

#### SECOND HARMONIC GENERATION EFFICIENCY MEASUREMENTS

The most widely used technique for confirming the SHG efficiency of NLO materials to identity the materials with non-centrosymmetric crystal structures is the Kurtz Powder technique. Fine powder of ZTS crystals were exposed under 1064nm laser beam from a pulsed Nd:YAG laser having a repetition rate of 10Hz and pulse width of 8ns to test the second harmonic generation (SHG) efficiency of the ZTS crystal was evaluated by taking the microcrystalline powder of KDP as the reference material. The output light was passed through a monochromator which detected green light at 532nm. This confirms the NLO behavior of the material. The relative conversion efficiency was calculated from the output power of ZTS crystals with reference to KDP crystals. It is observed that the conversion efficiency of ZTS is 0.8 times that of KDP crystal.

#### VI. CONCLUSION

Good quality single crystals of zinc thiourea sulphate were grown by the slow evaporation solution growth technique. The molecular structure was established by single crystal XRD analysis. The presence of functional groups was identified by FTIR spectral analysis. The UV-visible spectrum suggests the suitability of the grown crystal for various optical applications. The thermal stability of the title crystal was determined by TG/DTA studies. Vickers microhardness was calculated in order to understand the mechanical stability of the grown crystal. The powder SHG measurement shows that the grown zinc thiourea sulphate crystal has SHG efficiency.

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(An ISO 3297: 2007 Certified Organization)

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