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# Measurement of Dielectric Properties of Liquids Using Microwave Reflectometry Technique

Abdul Jaleel A. H<sup>1</sup>., A. R. Khan<sup>2</sup>, Maqdoom Faroqui<sup>3</sup> and Shaikh Y. H<sup>4</sup>.

Research Student Maulana Azad College, Aurangabad (MS) India<sup>1</sup>.

HOD, PG Dept of Computer Science, Maulana Azad College, Aurangabad (MS) India<sup>2</sup>

Principal, Maulana Azad College, Aurangabad (MS) India<sup>3</sup>.

Associate Prof. Shivaji College, Kannad, Aurangabad (MS) India<sup>4</sup>.

**ABSTRACT:** We present detailed experimental technique for the determination of dielectric constant ( $\epsilon$ ') and the dielectric loss ( $\epsilon$ '') for low loss liquids using microwave bench with plunger and coupler. The basic principle is the reflectometric analysis. The study presented here includes non-polar liquids are benzene and carbon tetrachloride. The dielectric constant and the dielectric loss of few selected liquids have been measured using microwave frequency 9.8 GHz and findings presented.

KEYWORDS: Microwave, Dielectric Constant, Dielectric Loss, Wave Guide Plunger, Reflectometry.

### I. INTRODUCTION

In relation to dielectric properties, the electromagnetic properties of material of interest to scientists and engineers include the permittivity, permeability and conductivity. For dielectric materials both for lossy and lossless, the most identifying parameter is the dielectric constant or permittivity. Dielectric properties are intrinsic characteristics of the materials explaining the behavior and degree of the wave-mater interaction when exposed to microwave field. These properties are very important in microwave heating, microwave sensing, process design and application. For example, there are many researchers who have used dielectric properties to measured moisture content of material and agro-food. A number of methods for measuring the microwave dielectric constant and dielectric loss of liquids have been proposed by different research groups and the classic paper on the method of measurement is that of Heston and Smythe Method for low loss liquids[1]. The measurement of dielectric properties of liquid materials at microwave frequencies can be used to determine some physical properties and evaluation of biological effects in biological molecules [3]-[5]. In this paper, the permittivity of lossless dielectric, benzene and carbon tetrachloride, low loss dielectric, aniline, ethyl alcohol and bromobenzene at X-band frequency of 9.8 GHz have been studied at room temperature (27°C) and results and findings are presented.

### **II. RELATED WORK**

The present work relates to the permittivity of standard non-polar liquid, benzene and carbon tetrachloride and aniline and ethyl alcohol have been measured at 9.8 GHz for mode TE10 and temperature of 27°C by using waveguide plunger technique. An X-band microwave bench was used to measure the guide wavelength, wavelength in dielectric, and a standing wave ratio. The block diagram of the experimental setup is shown in Fig. 1 and photograph of the experimental set up is shown in Fig. 2. In the present work we used two directional couplers with isolator for the measurement of the incident and reflected microwave in terms of current form the detector. This incident and reflected intensity of microwave was used for calculation of VSWR using the relation:

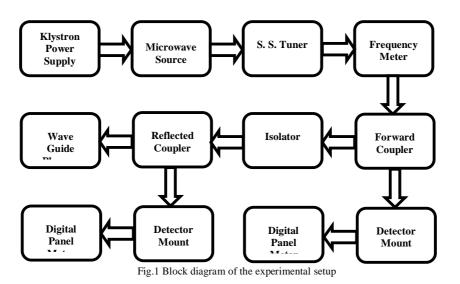


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$$VSWR = S = \frac{I_f + I_r + 2\sqrt{I_f I_r}}{I_f - I_r}$$



When microwave is travelling through medium 1 (air) and strikes normally to the surface of medium 2 (dielectric material), which one part of the energy is reflected, another part is transmitted through the surface and the rest of it is absorbed. The proportions of energy which fall into these three categories have been defined in term of the dielectric properties. The complex relative permittivity  $\varepsilon$  of the material is expressed as

$$\varepsilon = \varepsilon' - j \varepsilon''$$

(1)

Where,

 $\boldsymbol{\epsilon}'$  is the dielectric constant

 $\epsilon"$  is the dielectric loss

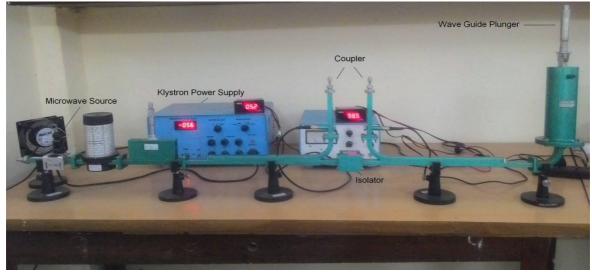


Fig 2 Photograph of experimental setup showing details of various components used like waveguide plunger and the two couplers along with the rest of the system.



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A standing wave pattern is thus produced in medium 1(Air). The input impedance Z(0)at the boundary is given by

$$Z(0) = Z_1 \left[ \frac{\rho - j \tan \frac{2\pi x_0}{\lambda_g}}{1 - j\rho \tan \frac{2\pi x_0}{\lambda_g}} \right]$$
(2)

Where  $x_0$  is the distance from a surface of dielectric material to the adjacent minimum of standing wave in medium 1(air) as shown in Fig.3 and  $Z_1$  is the impedance of medium 1.

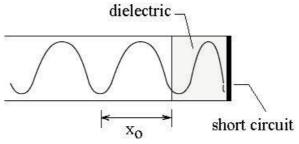


Fig.3 The distance of the first minimum for the interface  $(x_0)$ 

If medium 2 (dielectric under test) terminated in a short circuit, the input impedance of the interface is

$$Z_{(0)} = Z_2 \tanh \gamma_d d \tag{3}$$

Where  $Z_2$ ,  $\gamma_d$  and d are the characteristic impedance, propagation constant and the length of the medium 2, respectively, If we take liquid sample of length equal to the integral multiple of  $\lambda_d/2$  place in contact with the short circuit plate, then the distance of the first minimum for the interface ( $x_0$ ) is zero. Hence  $Z_1 = 1$ , so the equation (2) become

$$Z_{(0)SC} = \rho_n \tag{4}$$

 $\rho_n$  denotes inverse of VSWR when liquid length is  $n \frac{\lambda_d}{2}$ ,

If medium 2 is terminated in a short circuit the input impedance at the interface

$$Z_{(0)sc} = \rho_n = Z_2 \tanh \gamma_2 d \tag{5}$$

Since  $\gamma_2 = \alpha_d + j\beta_d$  and  $d = n\frac{\lambda_d}{2}$ , n being an integer, we have

$$\rho_n = Z_2 \tanh\left[\frac{n\alpha_d \lambda_d}{2}\right] \tag{6}$$

For low loss substances,  $tanh x \approx tan x \approx x$ ,

$$\rho_n = Z_2 \frac{n\alpha_d \lambda_d}{2} \tag{7}$$

We know that  $Z_2 = \lambda_d / \lambda_d$ . then equation (7) becomes

$$\rho_n = \frac{n\alpha_d \lambda_d^2}{2\lambda_g} \tag{8}$$



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However,  $\alpha_d$  used in equation (8) is the sum of two parts: one due to dielectric attenuation and the other due to plunger resistance and other losses. To eliminate later losses, a graph is plotted between several mean values of  $\rho_n$  versus n. The

slope of this graph gives the value of  $\frac{d\rho_{mean}}{dn}$ . The dielectric constant  $\varepsilon'$  and dielectric loss  $\varepsilon''$  are given by equation (1).

$$\varepsilon' = \left(\frac{\lambda_0}{\lambda_c}\right)^2 + \left(\frac{\lambda_0}{\lambda_d}\right)^2 \tag{9}$$
$$\varepsilon'' = \frac{2}{\pi} \left(\frac{\lambda_0}{\lambda_d}\right)^2 \left(\frac{\lambda_s}{\lambda_d}\right) \left(\frac{d\rho_{mean}}{dn}\right) \tag{10}$$

Where  $\lambda_0$  is free space wavelength,  $\lambda_d$  is wavelength in dielectric,  $\lambda_c$  is cut off wavelength = 2a. Here a is the width of the cell and  $\lambda_g$  is guide wavelength.

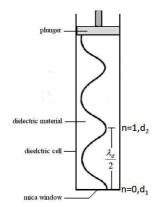


Fig.4 wavelength in dielectric

The relaxation time,  $\tau$  can be measured by

$$\tau = \frac{\varepsilon''}{\omega(\varepsilon' - n_D^2)}$$

Where  $n_D$  is the refractive index of the sample of dielectric and  $\omega = 2\pi f$ .

#### **II. EXPERIMENTAL RESULTS**

Table-1 The determined and literature values of dielectric constant, Dielectric loss and Relaxation time of solvents

Liquid	Experimental Value			Literature values		
	٤'	ε"	τ	<b>ε'</b>	ε"	τ
Benzene	2.56916	0.207706	4.3	2.26-2.28	-	4.12
Carbontetrachloride	2.522	0.200751	0.84	2.217	-	-
Acetone	20.661	6.635097	5.7	21.3		3.34
Aniline	5.809044	1.59745	3.7	5.597	1.448	-
Ethyl Alcohol	5.781161	0.840873	3.8	2.46	0.020	2.27
Acetonitrile	36.30878	4.161981	0.2	29.93533	4.595124	0.15350
Methanol	11.01672	9.8295	0.17	7.705522	4.335968	0.56271

(11)



(14)

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Three important dielectric properties i.e. Tangent loss (tan  $\delta$ ), penetration depth  $P_d$  and microwave conductivity  $\sigma$  are calculated using

Loss tangent = 
$$Tan\delta = \frac{\varepsilon''}{\varepsilon}$$
 (12)

Penetration Depth =

$$P_d = \frac{\lambda \sqrt{\varepsilon_r'}}{2\pi\varepsilon_r'} \tag{13}$$

Microwave Conductivity=  $\sigma = \frac{f\varepsilon''}{1.8 \times 10^{11}}$ 

for different liquids studied and are tabulated in Table - 2.

Parameter	Benzene	Carbon tetra chloride	Acetone	Aniline	Ethyl alcohol	Acetonitrile	Methanol
Tan ð	0.080846	0.079589	0.321141	0.15583	0.145451	0.114627	0.892239
P <sub>d</sub>	0.32469	0.32769	0.10746	0.20217	0.29465	0.08106	0.14716
σ	0.02491	0.10974	1.12947	0.04954	0.01614	0.22752	0.53735

Table-2 Experimentally determined values of solvents.

#### **III. CONCLUSION**

The results of measurement of dielectric properties shown above are in good agreement with the corresponding values given in literature. A slight deviation in few cases is attributed to the fact that the literature values shown correspond to different working conditions like temperature and particularly the microwave frequency[6,7] our working temperature was 27°C and microwave frequency was 9.8 GHz.

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#### REFERENCES

- [1] M.L. Sisodia, Bsic Microwave Techniques and Laboratory Manual. Newagepublishers, India, 2005.
- [2] Von Hippel, dielectric and Waves, The M.I.T. press, 1955.
- [3] W.Daseler, H.J.Steinhoff and A.Redhardt, A new method for the determination of the permittivity of small samples in the microwave range and its application to hemoglobin single crystals, Journal of Biochemical and Biophysical Methods, vol. 22, pp. 69-82,1991.
- [4] Y. Wang and M.N. Afsar, Measurement of complex permittivity of Liquids Using waveguide Techniques, Progress in electromagnetic Research, PIR, vol.42.pp 132-142,2003.

[5] D. Faktorova, Complex Permittivity of Biological Materials Measurement at Microwave Frequencies, Measurement Science Review, Vol.7, section 2, no.2,2007.
 [6] M.G. Corfield and et.al., Rapid method for determining v.h.f. dielectric parameters for liquids and solutions using standing wave procedures, British Journal of

- [6] And et al., Measurements on polar liquids at a microwave frequency evaluation of molecular parameters-a new method, Journal of Physics, vol.12, 1961.
  [7] M.B.R Murthy and et.al., Measurements on polar liquids at a microwave frequency evaluation of molecular parameters-a new method, Journal of Physics, vol.18,
  - M.B.R Murthy and et.al., Measurements on polar liquids at a microwave frequency evaluation of molecular parameters-a new method, Journal of Physics, vol.18, no.4, pp. 725-727,2003.
    M. Evaneform and teal. The Measurement of Dielectric Constant. Dinele Moment and Palayetian time of some liquid at Microwave Evanappear. Loss Journal on 1998.
  - [8] M.Fuangfoong and et.al., The Measurement of Dielectric Constant, Dipole Moment and Relaxation time of some liquid at Microwave Frequency, Laos Journal on Applied Science, vol.1, no.1, pp.197-200,2006.
  - [9] NBS, Tables of dielectric dispersion data for pure liquids and dilute solutions, National Bureau of Standards circular 589, U.S. Dept. of Commerce, Washington, DC.
  - [10] (S. B. Gedam and B. M. Suryavanshi , Dielectric Relaxation of Polar Molecules in BenzeneMedium at Microwave. International Journal of Applied Physics and Mathematics, Vol. 3, No. 4, July 2013(10)
  - [11] IA.P. Jogdand, 2 Dr. P.L. Kadam, Excess properties of acetonitrile + methanol binary mixtures at microwave frequency., IOSR Journal of Engineering (IOSRJEN) www.iosrjen.org ISSN (e): 2250-3021, ISSN (p): 2278-8719 Vol. 04, Issue 03 (March. 2014), ||V4|| PP 47-55
  - [12] A.P. Jogdand1, Dr. P.L. Kadam2, Dielectric behavior of acetonitrile + n-Butyl alcohol binary mixtures at microwave frequency., International Journal of Engineering Inventions e-ISSN: 2278-7461, p-ISSN: 2319-6491 Volume 4, Issue 3 (August 2014) PP:00-00.
  - [13] U. B. Tumberphale and et al. Dielectic relaxation study of binary mixtures of Aniline+ 2-Methaoxyethanol in Benzene from microwave absorption data.,Bionano Frontier, Vol.6 (1) Jan-June-2013.