

e-ISSN: 2319-8753 | p-ISSN: 2347-6710

EDIA

International Journal of Innovative Research in SCIENCE | ENGINEERING | TECHNOLOGY

INTERNATIONAL JOURNAL OF INNOVATIVE RESEARCH

IN SCIENCE | ENGINEERING | TECHNOLOGY

Volume 11, Issue 1, January 2022

INTERNATIONAL STANDARD SERIAL NUMBER INDIA

Impact Factor: 7.569

9940 572 462

6381 907 438

 \bigcirc



e-ISSN: 2319-8753, p-ISSN: 2347-6710 www.ijirset.com | Impact Factor: 7.569



|| Volume 11, Issue 1, January 2022 ||

DOI:10.15680/IJIRSET.2022.1101051

Eco Friendly Device for Carbon Dioxide Adsorption

Chetan Prasad S¹, Tejas M¹, Roshan Raj¹, Yogesh N¹, Dr.Prathibha B S², Shwethashree B³

U G Student, Department of Mechanical Engineering, B N M Institute of Technology, Bangalore, India¹

Associate Professor, Department of Chemistry, B N M Institute of Technology, Bangalore, India²

Assistant Professor, Department of Mechanical Engineering, B N M Institute of Technology, Bangalore, India³

ABSTRACT: Carbon dioxide is considered as a major contributor in the direction of global warming. The amount of carbon dioxide from automobiles is approximately sixty-fivepercent, which is higher than any other emission sources. Automobile vehicle engine exhaust design has become very important as exhaust emissions standards are ever increasing. There are challenging problems to the emission control designers majorly demanding high standards in exhaust gas conditions.

If we consider the upcoming stringent emission norms, this problem needs to be addressed appropriately. In this work, an attempt was made to capture CO_2 emission from automobile exhaust using Activated carbon. The concept of physical adsorption by Vander walls forces of attraction, Carbon di oxide found to be adsorbed on the surface of activated carbon. This device is an extension of the silencer. The device contains mainly bio degradable materials such as coconut coir, activated charcoal, corn cob etc.

Final results obtained shows that considerable amount of HC, CO, and CO_2 gets reduced. Activated carbon found to be a suitable one for CO_2 capture from engine exhaust and it has great scope to be implemented in a vehicle exhaust system.

KEYWORDS: Biomass, Biodegradable Materials, Carbon dioxide, Activated Charcoal

I. INTRODUCTION

In the current development, pollution is a word we get to hear often. Air pollution is a very dangerous form of pollution that affects environment as well as living organisms. Air pollution is caused by automobiles, transports, industrialization and in growing cities, which release harmful gases such as CO, NOx, HC, CO_2 causing a grave threat to the environment and living organisms.

Augmented use of cars in some major cities had resulted in serious apprehensions about urban air quality caused by engine exhaust gas emissions and the three main exhaust gas pollutants are hydrocarbons, carbon monoxide and nitrogen oxides. Controlling the emission from internal combustion engine can be accomplished by designing the composition of the fuel and better combustion is possible with improved engine design combustion technology for combustion of fuels. Engine modifications alone were not sufficient to control them, so catalytic systems were introduced. Catalytic converter plays an important role in eliminating the pollutants from engine exhaust.

II. LITERATURE REVIEW

Suhendrayatna et al., has observed that, activated carbon mass variables, concentration of activator and gas velocity at the exit will have influence on adsorption time and amount of CO gas adsorbed. He has observed through experiments that at optimum conditions 55.02 ppm of CO will be adsorbed in 1266 seconds[1]. Marcos et al., has used granular activated charcoal for adsorption of carbon di oxide at atmospheric and subatmoaspheric pressures which was prepared by using low cost biomass residue. It was observed that this adsorption was similar to chemically activated carbon using KOH[2]. Yuliusman et al., has used durian peel activated charcoal modified by using Magnesium oxide with the intension of increasing the adsorption capability. This was used to evaluate the adsorption of CO and HC exhaust gases and gave the best adsorption performance [3].

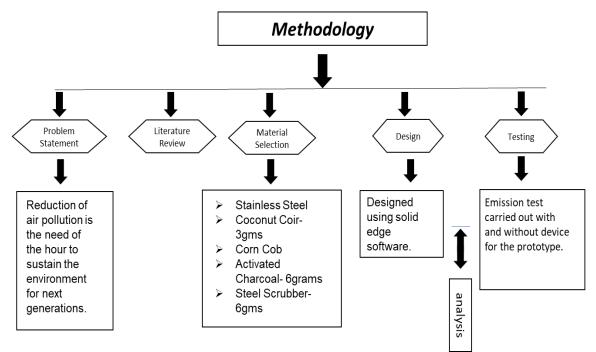
| e-ISSN: 2319-8753, p-ISSN: 2347-6710| <u>www.ijirset.com</u> | Impact Factor: 7.569|



Volume 11, Issue 1, January 2022

DOI:10.15680/IJIRSET.2022.1101051

III. METHODOLOGY



IV. SELECTION OF MATERIALS

Comparison between biomass activated charcoal-1 and Coconut Activated Charcoal.

Biomass activated charcoal-1	Coconut Shell			
It has more active centres	It has less active centres			
It has more adsorption capability	It has less adsorption capability			
It takes less time to enter into powder form	It takes more time to enter into powder form			

Comparison between Aluminum and Stainless Steel					
Aluminium	Stainless Steel				
Al has high strength to weight ratio but it is not	SS has less strength to weight ratio and it is stronger				
stronger.	than Al.				
Al is good conductor of heat and electricity	SS is poor conductor				
It has high thermal conductivity	It has less thermal conductivity				
Welding is difficult in Al	SS is easy to weld				
Al become soft at above 400 degrees temperature	SS can be used at much higher temperatures than Al				

International Journal of Innovative Research in Science, Engineering and Technology (IJIRSET)

e-ISSN: 2319-8753, p-ISSN: 2347-6710 www.ijirset.com | Impact Factor: 7.569



|| Volume 11, Issue 1, January 2022 ||

DOI:10.15680/IJIRSET.2022.1101051

Young's modulus is 70GPA	Young's modulus is 210GPA
Al is cheaper	SS is lightly costlier

• Comparison of Stainless Steel with Copper.

Stainless Steel	Copper		
SS has less strength to weight ratio and it is stronger than Al.	Cu has high strength to weight ratio and is less stronger than SS		
SS is made up of iron, chromium, nickel, manganese and copper. The chromium is added as an agent to provide corrosion resistance.	It has high a bit higher corrosion resistance than SS		
It has less thermal conductivity	It has high thermal conductivity		
SS is easy to weld	Copper is difficult to weld		
SS can be used at much higher temperatures.	Copper cannot be used at high temperatures.		
SS is poor conductor	It is good conductor of heat		
SS is lightly costlier	It is more expensive than SS		
Density of SS is 8*10^3 kg/m^3	Density of Cu is 8.96*10^3 Kg/cm^3		
Young's modulus is 210GPA	Young's modulus is 139GPA		

V. DESIGN DETAILS

- > The prototype consists of three parts mainly cap, body and the exhaust chamber.
- > All three parts are made using the Stainless-Steel material.
- > The device is an eight-inch stainless steel pipe that could be made to fit the exhaust pipe of the silencer.
- > The exhaust is provided with holes of dia 2mm using drill bit, for the exhaust gases to flow out.
- > The total length of the prototype is of 12cm and dia of body is 4cm and that of cap is 2cm.
- > The exhaust plate like structure is threaded internally so that it easily adheres to the body of the prototype.
- \blacktriangleright The dia of the exhaust pipe is of 4.5cm.
- > The entire body of the prototype is made up of Stainless-Steel metal.
- > The inner part of the prototype is filled with biodegradable materials which consists of
 - Coconut Coir = 2gms
 - Steel Scrubber = 4gms
 - Bio adsorbent = 10gms
- > The weight of the prototype weighs about 250 gms.
- > Prototype is designed by using Solid Edge Software. Prototype is shown in figure below

e-ISSN: 2319-8753, p-ISSN: 2347-6710 www.ijirset.com | Impact Factor: 7.569



|| Volume 11, Issue 1, January 2022 ||

DOI:10.15680/IJIRSET.2022.1101051



Fig:1 Prototype

VI. TESTING OF PROTOTYPE

To perform the Emission characteristics test, we made use of "Automotive Emission Analyzer (QRO-401 Series)" from Vemana Institute of Technology which is shown below.

	Est.	3(%)		-
• O REDMI NOTE 8 PRO CO AI QUAD CAMERA	CO 0 15 0 15 0 9 0 0 0 0 0 0 0 0 0 0 0	HC ppm 672 02 % AFR NOX	OROOTECH OROO-401 SELETT Image: Constant of the second o	

Fig: 2 Emission test

With the help of this analyzer, we were able to test three prototypes of different composition.

- Prototype 1: Metal scrubber + Without Activation charcoal + Coconut coir
- Prototype 2: Metal scrubber + (Bio massActivated charcoal-1) + Coconut coir
- Prototype 3: Metal Scrubber + (Bio mass Activated charcoal-2) + Coconut coir

International Journal of Innovative Research in Science, Engineering and Technology (IJIRSET)



| e-ISSN: 2319-8753, p-ISSN: 2347-6710| <u>www.ijirset.com</u> | Impact Factor: 7.569|

|| Volume 11, Issue 1, January 2022 ||

DOI:10.15680/IJIRSET.2022.1101051

VII. RESULTS AND DISCUSSION

Biomass Activated Charcoal-1

٠

Emission	Without Product	With Product	% Increase/De crease	After 250 kms	% Increase/D ecrease
НС	383ppm	242ppm	-36.81%	357ppm	-6.78%
CO ₂	9.6%	9.4%	-2.08%	4.2%	-56.25%
СО	0.72%	0.11%	-84.72%	0.09%	-87.5%
NO _X	46ppm	38ppm	-17.39%	34ppm	-26.08%
O ₂	6.24%	6.89%	+10.41%	13.33%	+53.18%

• Biomass Activated Charcoal-2

Emission	Without Product	With Product	% Increase/D ecrease	After 250 kms	% Increase/D ecrease
НС	677ppm	723ppm	+6.36%	633ppm	-6.5%
CO ₂	9.7%	7.7%	-20.61%	4.4%	-54.63%
СО	5.96%	5.33%	-10.57%	2.38%	-60%
NO _X	7ppm	13ppm	+46.15%	28ppm	+75%
O ₂	2.3%	4.72%	+51.27%	12.39%	+81.43%

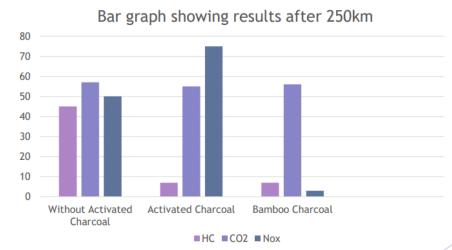
• Without activation charcoal

Emission	Without Product	With Product	% Increase/De crease	After 250 kms	% Increase/D ecrease
НС	595ppm	281ppm	-52.77%	322ppm	-45.88%
CO ₂	12.6%	9.6%	-23.8%	5.4%	-57.14%
СО	4.44%	3.21%	-27.70%	2.84%	-36.03%
NO _X	22ppm	31ppm	+29.03%	11ppm	-50%
O ₂	0.57%	3.93%	+85.5%	9.16%	+93.77%

e-ISSN: 2319-8753, p-ISSN: 2347-6710 www.ijirset.com | Impact Factor: 7.569

|| Volume 11, Issue 1, January 2022 ||

DOI:10.15680/LJIRSET.2022.1101051



From the experiment it is observed that biomass activated charcoal 2 has more adsorbing capacity compared to biomass activated charcoal 1 and wood activated charcoal. Because biomass activated charcoal 2 has more active sites compared to biomass activated charcoal 1 and wood activated charcoal.

VIII. CONCLUSION

In the course of time of the various phases of this project we have understood what it takes to model and build an actual working product. We conducted the tests initially for the prototype and later the same tests were performed on the device. What we observed from the above tests and experiments are across all the devices are:Decrease in the CO_2 emissions, Slight increase in NOx emissions, No drastic or significant changes in the bike's efficiency, i.e., mileage.

REFERENCES

1. "Optimization Process of Carbon Mono Oxide (CO) Gas Emission Adsorption using Activated Carbon from Rice Husk", Suhendrayatna, M Aprianti, P N Sheilla, I Machdar, Elvitriana, IOP Conf. Series: Journal of Physics: Conf. Series 1114 (2018) 012014

2. "Tailoring Low-Cost Granular Activated Carbons Intended for CO2 Adsorption", Marcos Juliano Prauchner, Silvia da Cunha Oliveira, Francisco Rodríguez-Reinoso, Frontiers in chemistry, November 2020, Volume 8, Article 581133

3. "Adsorption of Carbon Monoxide and Hydrocarbon Components in Motor Vehicle Exhaust Emission using Magnesium Oxide Loaded on Durian Peel Activated Carbon", Yuliusman, Maretha Putri Ayu, Afdhal Hanafi, Azmia Rizka Nafisah, Recent Progress on: Energy, Communities and Cities, AIP Conf. Proc. 2230, 030021-1–030021-6.

4."Characterization of Activated Carbon Fibres by CO2 Adsorption" D. Cazorla-Amoros, J. Alcañiz-Monge, and Linares-Solano Departamento de Química Inorgánica, Universidad de Alicante, Alicante, Spain.

5. Laboratory of Chemical Resource Engineering, Beijing University of Chemical Technology, Beijing 100029, People's Republic of China. "The role of different properties of activated carbon in CO2 adsorption" Guojun Yin, Zhenyu Qingya Liu, Weize Wu State Key.

6. "Emission reduction in automobile engine exhaust using bio-catalytic convertor: An Experimental Study", Rahul Ganji, Venkata Narayana Yenugula, Sreenidhi Institute of Science & Technology.

7. "CO2 adsorption on oxygen enriched porous carbon monoliths: Kinetics, isotherm and thermodynamic studies" Jasminder Singha, HaripadaBhuniab, SoumenBasua, School of Chemistry and Biochemistry, Thapar University, Patiala 147004, Punjab, India.

8. "CO2 adsorption in single-walled carbon nanotubes" Martin Cinke JingLi Charles W. BauschlicherJr.AlessandraRicca1M. Meiyappan NASA Ames Research Centre, Centre for Nanotechnology, MS 229-1, Moffett Field, CA 94035, USA.

9. "Control of Carbon Dioxide and other Emissions from Diesel Operated Engines using Activated Charcoal" - Shaik Sameer, Vijaya Balan P and Rajadurai MS Sharda Motor Industries Pvt Ltd, Mahindra World City, Department of Automobile Engineering, Hindustan Institute of Technology and Science, Hindustan University, Chennai, Tamil Nadu, India.

10. "Design and Fabrication of Eco-Friendly Aqua Silencer", S. Santhosh Sathish Kumar, G. Sarath Kumar, S. Saravana Manoj, M. Babu, Associate Professor Department of Mechanical Engineering Kings College of Engineering, Punalkulam.





Impact Factor: 7.569





INTERNATIONAL JOURNAL OF INNOVATIVE RESEARCH

IN SCIENCE | ENGINEERING | TECHNOLOGY





www.ijirset.com