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# Impact of CO<sub>2</sub> Concentration on its Growth and Pigment Formation of Anabaena Ambigua

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**ABSTRACT:** Due to fast growing concern about global warming, which is attributed primarily to the elevated  $CO_2$  level in atmosphere, photosynthetic  $CO_2$  mitigation has attracted much attention. The present study focuses on growth of Anabaena ambigua under different  $CO_2$  concentrations. Anabaena ambigua was grown in BG-11 media at a light intensity of 7000 lux in eight different bottles, arranged in columns. Each pair is named as control(0min), 2min, 5min, 8min respectively. Control indicates  $CO_2$  is not passed in first case,  $CO_2$  is passed for 2min twice daily with a time interval of 8 hrs at a flow rate of 3 Lpm in second case,  $CO_2$  is passed for 5min twice daily with a time interval of 8 hrs at a flow rate of 3 Lpm in third case,  $CO_2$  is passed for 8min twice daily with a time interval of 8 hrs at a flow rate of 3 Lpm in third case,  $CO_2$  is passed for 8min twice daily with a time interval of 8 hrs at a flow rate of 3 Lpm in third case,  $CO_2$  is passed for 8min twice daily with a time interval of 8 hrs at a flow rate of 3 Lpm in fourth case. Pigment analysis was performed. Phycobilin pigments are more in Anabaena ambigua grown in  $CO_2$  concentration of 2min, 5min, then control and lastly 8min. No much difference was observed in total chlorophyll content in the algal species grown at different CO2 concentrations. Carotenoids are more in algal cultures grown in  $CO_2$  concentrations of 2min, 5min than in control and finally in 8min

#### I. INTRODUCTION

Fossil fuels include oil, coal, natural gas. Dead plants, animals trapped beneath the earth's surface over a period of hundred's of millions of years produced fossil fuels(Ragauskas et al., 2006). Fuels consists of hydrocarbons which on burning releases energy. Main products obtained on burning the fossil fuels are carbon dioxides, nitrous oxides, and hydrocarbons (Veziroglu et al., 2008). These gases are referred as green house gases. These green house gases are heating the earth's atmosphere by trapping the infrared radiation which is otherwise reflected back towards space (Ramanathan, 1988, Schneider, 1989, and Green, et al., 2000). Among the gases emitted by fossil fuels carbon dioxide is one of the most significant. The earth's ecosystem provides a natural carbon cycle but humans are producing more amount of carbon dioxide from industries, automobiles which are the major sources for it (Benemann, 2001, and Govindjee et al., 1999). With risk of climate changes and inevitable fuel shortages scientists are focussing more on alternative energy sources as well as carbon dioxide sequestration(Ayhan Demirbas, 2008). Plants are the best source to remove the carbon dioxide from atmosphere. Many researchers are working on biodiesel, ethanol as the sources of fuel obtained from plants as burning of these fuels releases less pollutants. Different sources are identified like corn, soya bean, jatropha, microalgae etc(Pahl. Greg, 2008 and Knothe et al., 2005). The most important microalgae are diatoms, green algae, blue-green algae etc. Microalgae are photosynthetic microorganisms that convert sunlight, water and carbon dioxide to sugars, from which biological macromolecules, as lipids, can be obtained (Khan et al., 2009, GuanHua Huang et al., 2010, and Mandal et al., 2009). Photosynthetic carbon dioxide fixation by microalgae is a feasible technology with energy saving and environment friendly. In the applications of microalgae for carbon dioxide removal (CO<sub>2</sub> biofixation), and energy production, the tolerance to CO<sub>2</sub>, biomass, fatty acid content of microalgae are important(Teresa et al., 2010). Blue-green algae has wide applications in pharmaceutical industry, food industry, biofertilizers etc(Borowitzka, 1995, Spolaore et al., 2006,). It is also source of valuable products like pigments, polysaacharides etc(Rodgers et al., 1979).

The present work focusses on growth rates, pigments accumulation in Anabaena ambigua (blue-green algae) grown at different levels of carbon dioxide concentration, at light intensity of 7000 lux and in BG-11 media.

#### **II. MATERIALS AND METHODS**

The culture Anabaena ambigua (blue-green algae) used in this experiments was obtained from NCIM, Pune with Accession number 2785. The culture grown on solid agar media was subcultured in 150ml Erlenmeyer flasks This culture was grown photoautotrophically in the BG-11 media at 28°C under light and dark illumination (18:6) which was kept in the orbital shaker at 110 rpm for 12 days. Now sub-culture is available to perform further set of





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experiments. All the chemicals used in preparing the BG-11 media are of analytical grade purchased from Hi-Media. Distilled water is used in making BG-11 media

#### III. EXPERIMENTAL DESIGN

The subcultured Anabaena ambigua was inoculated into 8 fresh bottles containing 700 ml of BG-11 media in the laminar air flow chamber. The caps of the bottles were fitted with pipes to provide aeration for culture growth and were sealed with leak proof agent to prevent leakage of the media. The bottles were placed in an inverted position in a specific arrangement fitted with fluorescent bulbs with a light intensity of 7000 lux. The light intensity was measured using the lux meter. The bottles were placed in an upside down position within the arrangement and were fitted with aerators through the pipes for aeration to the cultures. Holes were made on the bottles of the bottles for the gases to escape. The culture in the media was allowed to grow for 10 days.

Now  $CO_2$  was supplied to the culture, growth was observed with change in colour in each bottle.  $CO_2$  was supplied from the cylinder to the arrangement as shown in the Fig1.



Fig 1: Growth of Anabaena ambigua at a light intensity of 7000 lux and different CO<sub>2</sub> Concentrations

As shown in the above Fig1 the bottles were arranged in four columns with two bottles in each column. The two bottles in each column represent the same conditions (i.e duplicates). Now  $CO_2$  was passed to the bottles twice a day with a time interval of 8hrs with a flow rate of 3 Lpm .  $CO_2$  was supplied till harvesting of biomass was done. Different concentrations of  $CO_2$  is supplied to the bottles as shown in the Table 1

Serial No(From left in Fig 1)	Carbon dioxide Concentrations supplied
First column of bottles	No carbon dioxide was passed. It is referred as control
Second column of bottles	CO <sub>2</sub> was passed for 2min twice a day with a time interval of 8hrs at a
	flow rate of 3 Lpm. It is referred as 2 min.
Third column of bottles	CO <sub>2</sub> was passed for 5min twice a day with a time interval of 8hrs at a
	flow rate of 3 Lpm. It is referred as 5min
Fourth column of bottles	CO <sub>2</sub> was passed for 8min twice a day with a time interval of 8hrs at a
	flow rate of 3 Lpm. It is referred as 8min

#### Table 1 Supply of CO<sub>2</sub> at different concentrations





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The above experiments were performed in duplicates from third day of inoculation the average absorbance values are recorded after passing the carbon dioxide gas, using colorimeter at 540nm. The recorded optical density values are shown in table 2.

**Biomass** extraction: After 10 days of growth the absorbance values found to be constant and then slowly decreased.CO<sub>2</sub> supply was stopped and the culture media was transferred into reagent bottles and was allowed to settle for 24 hours. When the particulate matter was found to be settled, the upper layers were removed by pipetting. To obtain the biomass, the matter was centrifuged at 6000 rpm for 6 minutes, pellets were obtained and the supernatant was discarded. The pellet was transferred into a petriplate and was allowed to dry in hot air oven at 60 degrees centigrade for 6 hours which gives the dry weight. The dry weight of the culture was noted.

#### **IV. RESULTS AND DISCUSSIONS**

The average absorbance values were recorded from 3<sup>rd</sup> day to 10<sup>th</sup> day and are given in table 2 as follows:

#### S.No Absorbance values at 540nm Control(0 min) 5 min 8 min 2 min 0.42 Day 3 0.62 0.46 0.63 0.69 0.78 0.59 0.53 Day 4 Day 5 0.79 0.9 0.68 0.57 1.02 0.74 0.64 Day 6 0.85 0.91 Day 7 1.05 0.83 0.71 Day 8 1.01 1.14 0.94 0.81 Day 9 1.07 1.16 0.92 0.86 Day 10 1.05 1.14 0.91 0.84

#### Table 2 (Absorbance values at different days)

The growth curves of Anabaena ambigua grown at different carbon dioxide concentrations was shown in the graph as follows:



Graph1: Growth curves of Anabaena ambigua at different CO<sub>2</sub> concentrations

The above graph indicates that growth was good in Anabaena ambigua where  $CO_2$  was supplied for 2min, then comes growth in control(0 min), followed by growth in 5min and 8min.





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#### Estimation of Pigments: Phycobiliproteins (Bennett and Bogorad, 1973)

Reagent: Phosphate buffer (0.05M. pH 7.5): Equal volumes of  $0.1M \text{ K}_2\text{HPO}_4$  and  $0.1M \text{ KH}_2\text{PO}_4$  were mixed to obtain 0.05 M phosphate buffer.

Procedure: A known volume (10 mL) of cyanobacterial cell suspension was tkenand centrifuged (4000g, 10 mins.) to obtain the pellet. Phycobilins were extracted completely from pellet with similar volume of 0.05 M phosphate buffer by repeated freezing and thawing. The absorbance was read at 562, 615 and 652 nm against phosphate buffer as blank. The phycobilins were calculated by using the following equations:

 $A_{652}-0.208 \times A_{615}$ Allophycocyanin (mg/mL) =  $\Box \sqcup \Box \sqcup \Box \sqcup \Box$ 5.06

 $A_{562}-[2.41 (PC)-0.849 (APC)]$ Phycoerythrin (mg/mL) =  $\Box \Box \Box$ 9.62

#### Table 4: Absorbance values at different CO<sub>2</sub> concentrations

Carbon dioxide	Absorbance values		
concentration	A562	A615	A652
Control	0.24	0.24	0.25
2 min	1.0	1.04	1.1
5 min	0.36	0.325	0.345
8 min	0.14	0.16	0.17

On substituting these absorbance values in the above equations we have summarized results as follows

i ubie et i ingeobilitio de diffei ene concente delono	Table 5:	<b>Phycobilins</b>	at different	concentrations
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Carbon dioxide	Phycobilins		
concentration	Phycocyanin(mg/ml)	Allophycocyanin(mg/ml)	Phycoerythrin(mg/ml)
Control			0.0227
2 min			0.0949
5 min			0.0346
8 min			0.0132

From the above table 5 we can conclude that the Phycobilin pigments are more in Anabaena ambigua grown in  $CO_2$  concentration of 2min, then 5min, then control and lastly 8min.

#### Chlorophyll (MacKinney, 1941)

(i) Reagent: Methanol (95%)

Procedure: A known volume (10 mL) of homogenized cyanobacterial suspensionwas taken and subjected to centrifugation (4000g, 10 mins.). The chlorophyll wasextracted from pellet with similar volume of methanol (95%) in a water bath (60<sup>o</sup>C, 30mins.). The suspension was centrifuged and the absorbance of the supernatant was read at650 and 665 nm against 95% methanol as blank.

The concentration of chlorophyll was calculated using the formula:

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Total chlorophyll (mg/mL) =  $2.55 \times 10^{-2} E_{650} + 0.4 \times 10^{-2} E_{665}$ Where,  $E_{650}$ = Absorbance at 650 nm

E<sub>665</sub>= Absorbance at 665 nm

#### Table 6 Chlorophyll absorbance at different CO2 concentrations

Carbondioxide concentration	Absorbance values	
	E650	E665
Control	0.147	0.304
2 min	0.151	0.288
5 min	0.1346	0.264
8 min	0.17	0.304

On substituting the above absorbance values we get the results as follows:

#### Table 7 Total chlorophyll at different CO2 concentrations

Carbon dioxide concentration	Total chlorophyll (mg/ml)
Control	0.495 x 10 <sup>-2</sup>
2 min	0.5 x 10 <sup>-2</sup>
5 min	0.448 x 10 <sup>-2</sup>
8 min	0.55 x 10 <sup>-2</sup>

From the above table 7 it was observed no much difference in total chlorophyll content in algal species grown in different CO<sub>2</sub> concentrations.

#### Carotenoids (Jensen, 1978)

#### Reagent: Acetone (85%)

Procedure: A known volume (10 mL) of homogenized cyanobacterial suspension

was taken and subjected to centrifugation (4000g, 10 mins.). Discarded the supernatant and washed the pellet 2-3 times with distilled water to remove traces of adhering salts. To the pellet, added 2-3 mL of acetone (85%) and stored at 4°C for a week. Centrifuged and collected the supernatant containing pigment. Repeated the extraction from pellet with acetone till the supernatant became colorless. Pooled all the fractions of supernatants and made up to a final known volume. Took the OD at 450nm using 85% acetone as blank and calculated the total amount of carotenoids as follows:

 $\begin{array}{c} D \ x \ V \ x \ f \ x 10 \\ Carotenoids \ (mg/mL) = \Box \Box \Box \Box \Box \Box \Box \Box \\ 2500 \end{array}$ 

Where, D = Absorbance at 450nm V = Volume of the extract, and f = Dilution factor

It is assumed that these pigments have an average extinction coefficient of 2500.

#### Table 8 Carotenoid absorbance at different CO<sub>2</sub> concentrations

Carbon dioxide concentration	D(Absorbance at 450nm)	V(Volume of the extract)	f(Dilution factor)
Control	0.3	10	10
2 min	0.5	10	10
5 min	0.53	10	10





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8 min	0.29	10	10

On substituting the above values in the equation we have the results as follows

Carbon dioxide concentration	Carotenoids (mg/ml)
Control	0.12
2 min	0.2
5 min	0.212
8 min	0.116

#### Table 9 Carotenoids at different CO2 concentrations

From the above table 9 it was observed that carotenoids are more in algal cultures grown in  $CO_2$  concentrations of 2min and 5min then in control and finally in 8min.

#### V. CONCLUSION

From the above experiments it can be concluded that growth was good in Anabaena ambigua where  $CO_2$  was supplied for 2min, then comes growth in control(0 min), followed by growth in 5min and 8min. Phycobilin pigments are more in Anabaena ambigua grown in  $CO_2$  concentration of 2min, 5min, then control and lastly 8min. Carotenoids are more in algal cultures grown in  $CO_2$  concentrations of 2min, 5min than in control and finally in 8min

#### REFERENCES

- 1. Ayhan Demirbas. Biodiesel. a realistic fuel alternative for diesel engines. Science and energy 2008 Turkey.
- Benemann, J. R. (2001). Wastewater treatment, greenhouse gas mitigation and hydrogen production using microalgae. In: K. Hiroyuki, & L. Yuan Kun (editors), Photosynthetic microorganisms in environmental biotechnology (2-9). Hong Kong: Springer
- Bennett, A. and L. Bogorad. 1973. Comparative chromatic adaptation in afilamentousblue-green alga. J. Cell. Biol. 58 :419–35.
- 4. Borowitzka MA, Microalgae as sources of pharmaceuticals and other biologically active compounds. J.Appl.Phycol,7: 3-15, (1995).
- 5. Green, Christopher. 2000. Potential scale-related problems in estimating the costs of CO2 mitigation policies. Climate Change 44: 331-349.
- Govindjee & Whitmarsh, J. (1999). The photosynthetis process. In: G. S. Singhal, R. Renger, K.-D. Sopory, Irrang, & Govindjee (editors), Concepts in photobiology: Photosynthesis and Photomorphogenesis (11-47). New Delhi: Narosa Publishing House.
- GuanHua Huang. Feng Chen. Dong Wei. XueWu Zhang and Gu Chen. Biodiesel production by microalgal biotechnology. Applied energy (2010) 87:38-46
- 8. Jensen, A. 1978 Chlorophylls and carotenoids. In: Handbook of Phycological methods, physiological and biochemical methods. J.A. Hellebust and J.S. Craige (edt.)pp.59-70 Cambridge University Press, Cambridge.
- Khan S.A., Rashmi, Mir Z.Hussain. Prospects of biodiesel production from microalgae in India. New Delhi. India 2009. Renewable and Sustainable Energy Reviews
- 10. Knothe A. and Van Gerpen J.H. The biodiesel Handbook. 2005Mac kinney G. absorption of light by chlorophyll solutions J.BIO.Chem., 1941
- 11. Mandal S., Nirupama Mallick, Microalgae Scenedesmus obliquus as a potential source for biodiesel production. Applied Microbiology Biotechnology (2009) 84:281-291
- 12. Pahl. Greg. Biodiesel: growing a new energy economy. 2nd Edition. Foreword by Bill McKibben. United States. 2008
- 13. Ramanathan, V., 1988. The greenhouse theory of climate change: a test by an inadvertent global experiment. Science 240: 293–299.
- 14. Ragauskas AJ, Williams CK, Davison BH, Britovsek G, Cairney J,Eckert CA, et al., The path forward for biofuels and biomaterials. Science **311**:484–489 (2006).





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- 15. Rodgers GA, Bergman B, Henriksson Eand Udris M. Utilization of blue-green algae as biofertilizers. Plant and soil, 52:99-107, (1979).
- 16. Spolaore P, Joannis-Cassan C, Duran E, Isambert A, Commercial applications of microalgae. J Biosci Bioeng, 101: 87-96,(2006).
- 17. Schneider, S. H., The greenhouse effect : science and policy. Science, 243 (1989) 771È81
- 18. Teresa M. Mata. Antonio A. Martins. Nidia S. Caetano. Microalgae for biodiesel production and other applications: A review. Renewable and Sustainable Energy Reviews (2010)14:217-232
- 19. Veziroglu, T. Nejat, and Sumer Shin. 2008. 21st Century's energy: Hydrogen energy system. Energy Conversion and Management 49: 1820-1831.









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